

Grain-Boundary Structure Influenced by Li-addition in Gd-Doped Ceria

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Abstract— High resistive ionic transport at the grain boundary is one of the most decisive factors for designing sophisticated nanostructured electrolytes as oxide ceramics. Here, the grain boundary structures are specifically investigated for nanostructured electrolytes: ceria (CeO_2), Gd-doped ceria (GDC), and GDC with 3 mol % and 9 mol % Li doping (Li-GDC), by means of positron lifetime spectroscopy. The influence of Li addition on the grain boundary structures is discussed on the basis of our recent model with the space charge layer.

Index Terms— GDC, CeO_2 , Gd_2O_3 , Li-GDC, grain boundary, positron, space charge layer.

1 INTRODUCTION

Nanostructured electrolytes have gained an interest for potential application as environmentally-friendly solid state fuel cells owing to high performance at intermediate temperatures (773 K – 973 K) [1]. In rare-earth-doped ceria, as e.g., Gd-doped ceria (GDC) [2], an ionic conductivity occurs by migration of oxygen ions through oxygen vacancies produced by the substitution of Ce^{4+} with Gd^{3+} . An availability of oxygen vacancies inside electrolytes is thus of significance for achieving the higher conductivity for oxygen ions [3]. It has been reported that the grain-boundary conductivity for rare-earth-doped ceria is lower than that of the bulk by several orders of magnitude [4]. The high resistive grain boundary for rare-earth doped ceria has been identified based on the results of impedance spectroscopy [4-5]. The impedance data are well explained by the space charge model, in which the grain boundary consists of a grain-boundary core and two adjacent space charge layers [4, 6-11]. In addition to that, cation phases have been often observed at the grain boundary due to dopant segregation by high-resolution transmission electron microscopy (TEM) [7, 12]. It has been thus believed that the high resistive grain boundary arises from the depletion of positively charged oxygen vacancies in the negative space charge layers [7].

A positron, anti-particle of electron, is a positively charged particle that can sensitively localize at the negatively charged spaces on an angstrom scale in materials. In metals, intermetallic compounds [13] and quasicrystals [14], satisfactory results have been given for the studies on lattice vacancies that are negatively charged relevant to the matrix due to missing of positive ion core. It has been demonstrated that the negatively charged part localized on polar elements in polymers is probed by positrons [15-16]. The size of open space can be investigated through the measurements of positron lifetimes by positron lifetime spectroscopy [13-16]. In the present study, the grain-boundary structures were studied for CeO_2 and GDC on an atomic scale by positron lifetime spectroscopy.

GDC doped with different amount of Li (Li-GDC) was examined as well since Li is one of the effective sintering aids due to the capillary effect of liquid phase into the grain boundaries [18-19]. An influence of Li addition on the grain boundary structures is discussed on the basis of our recent model with the space charge layer [20].

2 EXPERIMENTS

GDC was prepared using oxalate coprecipitation method [21-22]. The Ce-Gd nitrate solution mixed at a Gd molar ratio of ~ 0.2 was dropped into stirred oxalate acid solution to produce oxalate precipitates. The precipitates were calcined at 873 K for 1 h under the atmospheric condition to form oxides. Li-GDC was prepared by adding LiNO_3 into GDC with ethanol as dispersant so that 3 mol % and 9 mol % Li can be obtained, which are referred as Li-GDC1 and Li-GDC2, respectively, thereafter. The powder samples were compacted into pellets by uniaxial pressing with a pressure of 20 MPa and sintered at 873 K for 1 h in the air. The pellets of CeO_2 and Gd_2O_3 were prepared with the same procedure as that of GDC and sintered at 1073 K for 2 h in the air.

Positron lifetime spectroscopy was performed at room temperature. The positron source (^{22}Na), sealed in a thin foil of Kapton, was mounted in a sample-source-sample sandwich for the measurements. Positron lifetime spectra ($\sim 1 \times 10^6$ coincidence counts) were recorded with digital oscilloscope-based system, in which the time resolution of 190 ps full-width at half-maximum (FWHM) was achieved. The lifetime spectra were numerically analyzed using the POSITRONFIT code [23].

3 RESULTS AND DISCUSSION

Table 1 lists the results of positron lifetime spectroscopy for CeO_2 , GDC, Li-GDC1, and Li-GDC2 together with those of Gd_2O_3 and yttria-stabilized zirconia (s-YSZ) single crystal, which are similar materials of oxide family. In contrast with YSZ single crystal where a single component of positron lifetime corresponding to annihilation in the matrix is observed, two components of positron lifetime are obtained for CeO_2 , GDC, Li-GDC1, Li-GDC2, and Gd_2O_3 . The average positron

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diffusion length in the YSZ single crystal without any trapping center is reported as $L_+ \approx 60$ nm [24]. On the other hand, the present samples of CeO₂, GDC, and Li-GDC are polycrystalline materials that consist of the particle and the crystallite of minimum structural unit with structural perfection, as discussed in detail in our earlier paper [22]. As revealed by XRD experiments, the crystallite sizes of CeO₂, GDC, and Li-GDC are 18 nm, 14 nm, and 27 nm [20], which are by far smaller than the positron diffusion length in YSZ single crystal. Positrons implanted in the crystallites of CeO₂, GDC, Li-GDC, and Gd₂O₃ are thus expected to diffuse sufficiently to reach the grain boundaries, indicating that the positron annihilation sites are exclusively located at the grain boundaries. Besides the present GDC and Li-GDC, positron annihilation at the grain boundaries have been often reported for nanocrystal-embedded amorphous [14] and nanometer-sized polycrystal [26] so far.

TABLE 1
POSITRON LIFETIMES WITH THEIR RELATIVE INTENSITIES OBTAINED FOR CeO₂, GDC, AND TWO KINDS OF LI-GDC. THE DATA FOR Gd₂O₃ AND YSZ SINGLE CRYSTAL ARE PRESENTED FOR COMPARISON.

Sample	τ_1 [ps]	I_1 [%]	τ_2 [ps]	I_2 [%]	τ_3 [ps]	I_3 [%]
CeO ₂	183±2	90±2			441±6	10±2
Gd ₂ O ₃	193±3	72±2			551±6	28±2
s-YSZ	180±2	100				
GDC			253±4	77±2	501±6	23±2
Li-GDC			225±3	87±2	332±5	13±2
Li-GDC2			214±4	92±2	323±5	7±2

The shorter lifetimes τ_1 of dominant components obtained for CeO₂ and Gd₂O₃ are close to that of YSZ single crystal, signifying the presence of tightly packed region as the grain-to-grain contact essentially identical to bulk. On the other hand, significantly longer lifetimes τ_2 are obtained as dominant components for GDC, Li-GDC1, and Li-GDC2, which are attributable to positron annihilation in the space with the size of a few atoms. Further longer positron lifetimes τ_3 for CeO₂, GDC, Li-GDC1, Li-GDC2 and Gd₂O₃ can be ascribed to the void-like open spaces created by lattice mismatching at the grain boundaries. The positron lifetime of 441 ps for CeO₂ increases up to 501 ps for GDC, indicating that the void-like open spaces at the grain boundaries are enlarged upon Gd³⁺ doping into CeO₂. The positron lifetime of 501 ps for GDC decreases down to 332 ps for Li-GDC1, indicating the shrinkage of void-like open spaces upon Li⁺ doping into GDC. This shrinkage effect is further enhanced with increasing Li addition up to 9 mol %. In the light of the fact that both the values of positron lifetime τ_3 and its relative intensity I_3 for Li-GDC are smaller than those of GDC, an addition of Li significantly densifies the grain boundaries shrinking the void-like open spaces.

Positron is known to act as a highly sensitive probe of the angstrom-scale local spaces with negative charges. The observed annihilation spaces with the size of a few atoms located

at the grain boundaries of GDC and Li-GDC are negatively charged. As clarified by our previous coincident Doppler broadening spectroscopy, the negatively charged spaces of GDC and Li-GDC are associated with Gd and Li at the grain boundaries, respectively. Our former study has further indicated that a small amount of lattice cation Gd³⁺ with below the detection limit of XRD is heterogeneously distributed at the grain boundaries even after calcination [22]. The segregation of Gd³⁺ to the grain boundaries creates positively charged local spaces, by which negatively charged spaces adjacent to them are formed. The cation of Li⁺ added as sintering aid segregates to the grain boundaries being positively charged local spaces as observed by TEM earlier [19], which forms the negatively charged spaces adjacently. The higher relative intensity I_2 for Li-GDC1 and Li-GDC2 than that of GDC indicates that the negative charge for Li-GDC is more densely distributed than that of GDC.

The grain-boundary structures of CeO₂, GDC, and Li-GDC have been modeled based on the results of positron lifetime and coincident Doppler broadening spectroscopy [20], as schematically shown in Fig. 1. The grain boundary of CeO₂ is composed of grain-to-grain contact and void-like open space, which are denoted as local atomic sites a and c, respectively. Upon Gd doping into CeO₂, the grain-boundary width expands owing to segregation of Gd³⁺, disappearing the grain-to-grain contact and thus enlarging void-like open space. An addition of 3 mol % Li into GDC as sintering aid causes segregation of Li⁺ into the grain boundary and narrows the width of grain boundary, shrinking the void-like open spaces. This effect could be enhanced by increasing Li addition up to 9 mol %. Segregation of Gd³⁺ and Li⁺ creates negatively charged spaces with the size of few atoms adjacently to the segregated cations, denoted as local atomic site b. These negatively charged spaces identified here are the origin of space charge layer, which has been discussed in the context of space charge theory for the grain-boundary conductivity.

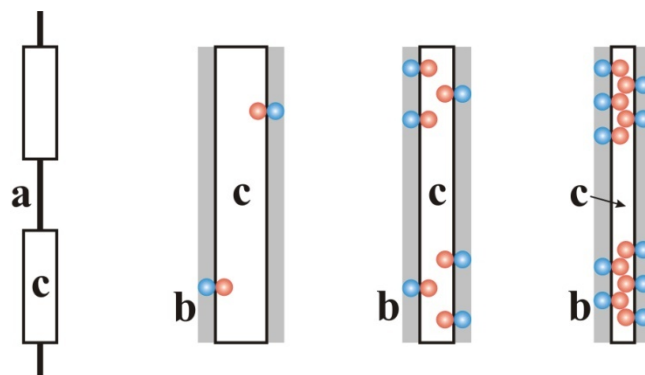


Fig. 1. Schematic illustrations of grain-boundary structures for CeO₂, GDC, and Li-GDC. The local atomic sites a, b, and c are identified by the present positron annihilation spectroscopy. The local atomic sites a and c correspond to grain-to-grain contact and void-like open space created by lattice mismatching, respectively. Red circles are positively charged local spaces caused by Gd³⁺ and Li⁺ segregates. Blue circles denoted as b are negatively charged spaces adjacent to the positive charges. Gray shadows are negative-space charge layers originated from the negatively charged spaces b.

4 CONCLUSION

The grain-boundary structures were studied for CeO₂, GDC, and Li-GDC by means of positron lifetime spectroscopy. The void-like open spaces created by lattice mismatching in addition to the grain-to-grain contacts essentially identical to the bulk were found at the grain boundaries for CeO₂. An addition of Li⁺ as sintering aid significantly densifies the grain boundaries. This effect is further enhanced by an increase of Li addition. Both the cations of Gd³⁺ and Li⁺ segregate to the grain boundaries decreasing the grain-to-grain contacts and forming positively charged local spaces for GDC and Li-GDC, by which the negatively charged spaces adjacent to them are created. The angstrom-scale spaces with negative charge identified in the present study are the origin of space charge layer, which has been discussed in the context of space charge theory for the grain-boundary conductivity.

ACKNOWLEDGMENT

The author is indebted to T. Kosaka (Tokyo Gakugei University) for supplying GDC samples. This work was partially supported by a Grant-in-Aid of the Japanese Ministry of Education, Science, Sports and Culture (Grant Nos. 25400318 and 25400319).

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